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Ahmed Attayibat,^a Smaail Radi,^a Abdelkrim Ramdani,^a Driss Eddike,^b Monique Tillard^c and Claude Belin^c*

^aLaboratory of Physical-Organic Chemistry, Department of Chemistry, Faculty of Sciences, University Mohammed The First, BP 524, 60 000 Oujda, Morocco, ^bLaboratory of Inorganic Solid Chemistry, Department of Chemistry, Faculty of Sciences, University Mohammed The First, BP 524, 60 000 Oujda, Morocco, and ^cLaboratoire des Agrégats Moléculaires et Matériaux Inorganiques, UMR 5072 CC15, Université de Montpellier II, Sciences et Techniques du Languedoc, 2 Place Eugène Bataillon, 34095 Montpellier Cédex 5, France

Correspondence e-mail: eddike@sciences.univ-oujda.ac.ma

Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.003 Å R factor = 0.049 wR factor = 0.125 Data-to-parameter ratio = 14.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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2-(1',5',5-Trimethyl-3,3'-bi-1H-pyrazol-1-yl)ethanol

In the crystal structure of the title compound, $C_{11}H_{16}N_4O$, the two pyrazole rings are coplanar, as indicated by both the dihedral angle of 0.8 (1)° between their mean planes and the N-C-C-N torsion angle of 179.81 (13)°.

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Comment

Pyrazole derivatives are of particular interest in coordination chemistry. Several pyrazole-containing compounds have shown ability to complex transition metal ions (Jeong *et al.*, 2003; Bol *et al.*, 1997; Mary *et al.*, 1993; Elguero *et al.*, 1980). However, the literature reports few applications of *C*,*C*bipyrazole derivatives in coordination chemistry. In this paper, we have synthesized and determined the X-ray crystal structure of a new *C*,*C*-bipyrazole ligand, (1',5',5-trimethyl-3,3'-bi-1*H*-pyrazol-1-yl)ethanol, (II), functionalized with a mobile chain containing a donor heteroatom. Ligand (II) was prepared by reduction of (1',5',5-trimethyl-3,3'-bi-1*H*-pyrazol-1-yl)ethyl acetate, (I), by LiAlH₄; it has two pyrazole Nsp² atoms and one O atom at the end of a flexible side chain.



The pyrazole-to-methyl bond lengths C11–C12 and C5–C6 (Table 1) are very close to those observed in 1-(*p*-nitrophenyl)-3-methyl-4-bromopyrazole (Lapasset & Falgueirettes, 1972). The N1–N2 and N3–N4 bond lengths are within the values [1.2348 (8)–1.385 (4) Å] reported for N–N bonds in pyrazole rings (Krishna *et al.*, 1999).

The atoms of the pyrazole rings do not deviate from their least-squares mean planes by more than 0.0018 (9) Å (atoms N1 and C8). These deviations may be compared with that (0.012 Å for C6) in 1-(*p*-nitrophenyl)-3-methyl-4-bromopyrazole (Lapasset & Falgueirettes, 1972).

With a N1–C8–C9–N3 torsion angle of 179.81 (13)°, the two heterocycles are coplanar and their N atoms are *trans* with respect to the C8–C9 bond. This configuration has already been observed in the 3,3'-bipyrazole (Monge *et al.*, 1994). Electron delocalization and conjugation in the pyrazole π systems may occur between the two rings. This is indicated by the shortening of the C8–C9 bond length to 1.463 (2) Å [1.4586 Å in the 3,3'-bipyrazole (Monge *et al.*, 1994)] and also by the small dihedral angle of 0.8 (1)° between the pyrazole mean planes. The low steric hindrance between substituents at positions 1 (or 1') and 5 (or 5') results in small values of 2.7 (2) and 0.4 (2)°, respectively, for the C13-N4-C11-C12 and C6-C5-N2-C3 torsion angles.

In the crystal structure, molecules of (II) form hydrogenbonded centrosymmetric dimers through $O-H\cdots N$ bonds (Table 2).

Experimental

Bipyrazole (I) (0.63 g, 2.40 mmol) dissolved in tetrahydrofuran (THF, 20 ml) was added to a stirred suspension of LiAlH₄ (200 mg, 5.26 mmol) in THF (10 ml) cooled to 273 K. The mixture was stirred under reflux for 2 h. After cooling to 273 K, the excess hydride was destroyed by slow addition of water (0.2 ml), aqueous sodium hydroxide (15%, 0.2 ml) and then water (0.6 ml). The stirring was continued for 2 h, the solid material filtered off and the residue washed with hot THF. The filtrate and THF washings were concentrated under reduced pressure and the resulting solid was recrystallized from THF as colourless crystals of (II) suitable for X-ray analysis (yield 72%; m.p. 451-453 K).

 $V = 549.2 (1) \text{ Å}^3$

 $D_{\rm r} = 1.332 {\rm Mg} {\rm m}^{-3}$

Parallelepiped, colourless $0.35 \times 0.30 \times 0.12 \text{ mm}$

2185 independent reflections

1797 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation $\mu = 0.09 \text{ mm}^{-1}$

T = 173 (2) K

 $\begin{aligned} R_{\rm int} &= 0.077\\ \theta_{\rm max} &= 26.1^\circ \end{aligned}$

Z = 2

Crystal data

$C_{11}H_{16}N_4O$
$M_r = 220.28$
Triclinic, $P\overline{1}$
a = 6.9135 (7) Å
b = 7.3389 (8) Å
c = 12.183 (1) Å
$\alpha = 86.09 \ (1)^{\circ}$
$\beta = 82.18 \ (1)^{\circ}$
$\gamma = 63.75 \ (1)^{\circ}$

Data collection

Oxford Diffraction Xcalibur CCD
diffractometer
ω scans
Absorption correction: none
7672 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0485P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.049$	+ 0.2622P]
$wR(F^2) = 0.125$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.06	$(\Delta/\sigma)_{\rm max} < 0.001$
2185 reflections	$\Delta \rho_{\rm max} = 0.20 \text{ e} \text{ \AA}^{-3}$
152 parameters	$\Delta \rho_{\rm min} = -0.27 \text{ e} \text{ Å}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Selected geometric parameters (Å, °).

N1-N2	1.3650 (18)	N4-C13	1.455 (2)
N2-C3	1.460 (2)	C12-C11	1.494 (2)
N3-N4	1.3595 (19)	C5-C6	1.491 (2)
C8-C9	1.463 (2)		
N1-C8-C9-N3	179.81 (13)	C3-N2-C5-C6	0.4 (2)
N3-N4-C11-C12	-178.96 (13)		



Figure 1

View of (II). Displacement ellipsoids for the non-H atoms are drawn at the 30% probability level.

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O-H\cdots N1^i$	0.87 (3)	2.10 (3)	2.9441 (19)	162 (2)

Symmetry code: (i) -x, -y + 1, -z + 1.

Except for the hydroxy H atom, which was located in a Fourier difference map and whose position and isotropic displacement parameter were freely refined, all H atoms were placed in geometrically idealized positions (C-H = 0.93–0.97 Å) and constrained to ride on their parent atoms, with $U_{\rm iso}(\rm H) = 1.5U_{eq}(\rm C)$ for methyl groups and $U_{\rm iso}(\rm H) = 1.2U_{eq}(\rm C)$ for other H atoms.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2004); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2004); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Version 1.08; Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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